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E1-26 The room volume is  $(21\ 13\ 12)\text{ft}^3(0.3048\ \text{m}/\text{ft})^3 = 92.8\text{m}^3$ . The mass contained in the room is  $(92.8\text{m}^3)(1.21\text{kg}/\text{m}^3) = 112\text{kg}$ : E1-27 One mole of sugar cubes would have a volume of  $N_A \cdot 1.0\ \text{cm}^3$ , where  $N_A$  is the Avogadro constant. Since the volume of a cube is equal to the length cubed,  $V = l^3$ , then  $l = \sqrt[3]{N_A\ \text{cm}^3} = 8.4 \cdot 10^7\ \text{cm}$ . E1-28 The number of seconds in a week is

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