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Modern Chemistry Chapter 7 Quiz

Modern Chemistry Chapter 7 test. STUDY. PLAY. The basis of an ionic bond is the__. absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a__. noble gas. Ions that are formed from more than one atom are called__.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: True a. If the formula mass of one molecule is x amu, the molar mass is x g/mol. False b. Samples of equal numbers of moles of two different chemicals

7 Chemical Formulas and Chemical Compounds

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ONLINE Chemistry HMDScience.com ONLINE LABS include: Test for Fe(II) and Fe(III) Water of Hydration Determining the Empirical Formula of Magnesium Oxide ... CHAPTER 7 Untitled-2 206 6/28/2011 2:09:06 PM. C 8H 18 Subscript indicates that there are 8 carbon atoms in a molecule of octane.

CHAPTER 7 Chemical Formulas and Chemical Compounds

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Modern Chemistry Chapter 7 Answer Key

Quiz Modern Chemistry 33 . Name Section Quiz, continued Class Date 6. The Lewis structure of which of the following compounds does not follow the octet rule? a. NH_3 b BF_3 c. d. H_2O 7. In a double covalent bond, a. one atom has more than eight valence electrons.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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